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In vitro Antioxidant Activity and Total Polyphenols Content of Wild Edible Polypore Mushrooms – Bondazewia berkeleyi and Ganoderma lucidum

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Authors' contributions

This work was carried out in collaboration among all authors. Author EEE designed the study, wrote the protocol, and the first draft of the manuscript. Authors BIU and NSP conducted experimental work, managed the analyses of the study and literature searches. All authors read and approved the final manuscript.

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Original Research Article

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ABSTRACT

Aim: To determine the total phenols content and antioxidant activity of *B. berkeleyi* and *G. lucidum* methanol extracts.

Study Design: *In vitro* evaluation of antioxidant assays and quantitative determination of total phenolics and flavonoids content of *B. berkeleyi* and *G. lucidum* mushroom extracts.

Place and Duration of Study: Department of Chemistry, University of Uyo, Nigeria (August – November, 2014).

Methodology: The total phenolics content was estimated using Folin-Ciocalteu method and total flavonoid content was determined using aluminum chloride. *In vitro* antioxidant activities of extracts were studied in different systems including 2, 2-diphenyl-1-picrylhydrazyl (DPPH) scavenging assay and metal chelating activity along with standards.

Results: Total phenols content in *B. berkeleyi* and *G. lucidum* was 32.99 and 42.7 mg GAE/g and flavonoids amounted to 22.4 and 35.23 mg QE/g respectively. The extracts of *B. berkeleyi* and *G. lucidum* exhibited significant DPPH radical scavenging activity with IC_{50} values 22.9 µg/ml and 14.6 µg/ml; metal chelating capacity of 67.4% and 82.6% at 500 µg/mL respectively.

Conclusion: The results of this study substantiates the use of the extract of *G. lucidum* powder as an antioxidant agent and serves as a pointer for further exploitation of *B. berkeleyi* mushroom for its antioxidant compounds and other potential biological activity.

Keywords: Macrofungi polypore; Bondazewia berkeleyi; Ganoderma lucidum; antioxidant activity.

1. INTRODUCTION

Mvconutraceuticals are neutraceutical substances derived from fungi. Natural products have been the source of most of the active ingredients of medicines [1]. Macrofungi have long been used as valuable food source and as traditional medicines around the world [2]; they are known to produce large and diverse variety of secondary metabolites [3]. These secondary metabolites have health promoting properties such as antioxidant, antimicrobial, anticancer, cholesterol lowering and immunostimulatory effects [4-6]. Antioxidant compounds have important ability to trap free radicals and thus inhibit the oxidative mechanisms that lead to degenerative diseases. Naturally occurring antioxidants can be found in whole grains, fruits, vegetables, teas, spices and herbs. Mushrooms have also been reported as organisms with antioxidant activity which is correlated with their phenolic and polysaccharide compounds [7,8].

The fungus B. berkeleyi (Fr.) Bond. et singer (synonymous: Polyporus berkeleyi) of the family Bondarzewiaceae (Basidiomycota) grows at the base or roots of Abies and other conifers of the family Fagaceae. It is edible when young and gets tough and unappetizing with age [3]. Bernd et al. [9] reported a cytotoxic metabolite, montadial A, isolated from the polypore B. montana. Reishi mushroom, G. lucidum (Fr.) Krast (Polyporaceae) is used for the treatment of bronchitis, hepatitis, hypertension, tumorigenic diseases and immunological disorders [10] and as remedy for the promotion of health and longevity [11]. A number of polysaccharides have been isolated, characterized and assayed for antiturmor, antimetastatic and immunological activity [12,13]. Ganoderma lucidum, is utilized as dried whole, powder or capsule and as tablets for promoting health in humans [14] and in some countries like China and Japan it is cultivated and utilized as source of feed supplement [15,16]. However, the pharmacological efficacy and bioactive components in Ganoderma from different origins or hosts are significantly different [17,18] and the reasons leading to this phenomenon are still unclear. There are many synthetic antioxidants butylated like

hydroxytoluene (BHT) and butylated hydroxyanisole (BHA) which are commonly used, but they are reported to have side effects and are carcinogenic [19]. Therefore, there is an increased interest in the use of natural antioxidants due to their presumed safety, nutritional and therapeutic value [20]. This may explain the interest in examining plants and mushroom extracts as sources of cheaper and effective antioxidants and the growing interest in nutraceuticals. In addition, there seems to be no report hitherto on the antioxidant activity of B. berkeleyi and a paucity of information on the antioxidant potential of G. lucidum collected in Nigeria. Therefore, in continuation of our work on the nutraceutical potential of mushroom polypores [21], this study seeks to assess the total polyphenols content and antioxidant activity of B. berkeleyi and G. lucidum.

2. MATERIALS AND METHODS

2.1 Samples Collection and Extraction

The fresh sporocarps of mature mushroom species (B. berkeleyi and G. lucidum) were collected from the forest in the rainy season (July to August) in Uyo and Ikono Local Government Areas of Akwa Ibom State. Specimen identifications and authentication were done by a mycologist, Dr. Joseph Essien and the voucher specimens were deposited in the School of Pharmacy herbarium, University of Uyo, Nigeria. The macroscopic descriptions, including size, shape, color, texture, and odor, were noted. The color of the carpophore, shape of the cap and stipe, color of the flesh and latex, and its smell and habitat were also noted. The mushroom samples were packed in opaque plastic bags and transferred to the analytical laboratory. The samples were carefully cleaned manually to remove any extraneous materials, cut, sun-dried and oven-dried (Gallenkamp, DV 333) at 45°C for 40 h to constant weight. Dried samples were pulverized using an agate homogenizer, and stored in pre-cleaned polyethylene bottles, prior to analyses. All reagents were of analytical grade, except otherwise stated. The ground samples were extracted with methanol using a

Soxhlet apparatus. The extract was concentrated under vacuum using a rotator evaporator.

2.2 Determination of Total Phenolics

The amount of total phenols in the methanol extracts *B. berkeleyi* and *G. lucidum* was determined by Folin-Ciocalteu's reagent method [22]. The extracts (0.5 ml) and Folin-Ciocalteu's reagent (0.5 N, 0.1 ml) were mixed and the mixture was incubated at room temperature for 15 min. Then saturated sodium carbonate solution (2.5 ml) was added and further incubated for 30 min at room temperature. The absorbance was measured at 760 nm. Gallic acid was used as a positive control. Total phenol values are expressed in terms of gallic acid equivalent (mg/g of extract).

2.3 Determination of Flavonoids Content

The flavonoids content in the mushroom extracts was determined by aluminium chloride colorimetric method [23]. The reaction mixture (3 ml) consisted of sample (1 ml, 1 mg/ml), aluminium chloride (1.2%, 0.5 ml) and potassium acetate (120 mM, 0.5 ml) was incubated at room temperature for 30 min. The absorbance of all samples was measured at 415 nm. Quercetin was used as positive control. The flavonoid content is expressed in terms of quercetin equivalent (mg/g of extract).

2.4 DPPH Radical Scavenging Activity

DPPH radical scavenging activity of each extract was determined according to the method of Blois [24]. DPPH (0.1 mM) in methanol was prepared and the solution (1 ml) was mixed with crude extracts (1.0 ml) prepared in methanol at different concentrations (20, 40, 60, 80, and 100 μ g/ mL). The mixture was shaken and kept for 30 mins at room temperature. The decrease of solution absorbance due to proton donating activity of components of each extract was determined at 517 nm. BHA was used as the positive control. The DPPH radical scavenging activity was calculated using the following formula:

% inhibition =
$$\frac{A_{control} - A_{sample}}{A_{control}}X$$
 100

2.5 Metal Chelating Activity

The method of Dinis et al. [25] was used. Crude extract (0.5 g) was mixed with FeCl_2 (2 mM, 0.05 ml) and Ferrozine (5 mM, 0.4 ml). The total volume was diluted with methanol (2 ml). The mixture was shaken vigorously and left standing at room temperature for 10 mins. After the mixture had reached equilibrium, the absorbance of the solution was measured at 562 nm in a spectrophotometer. The percentage inhibition of ferrozine Fe²⁺ complex was calculated using the formula:

% inhibition of ferrozine – Fe2+=
$$\frac{A_{control} - A_{sample}}{A_{control}} X$$
 100

3. RESULTS AND DISCUSSION

The results of total phenolics content and flavonoids are given in Table 1. Ganoderma lucidum methanol extract was found to contain higher amounts of polyphenols and flavonoids compared with B. berkeleyi extract. The estimated polyphenols content for G. lucidum extract in this study is similar to total phenolics (71 µg GAE/mg) and flavonoids (45 µg QE/mg) content of the Indian sample, G. applanatum reported by Nagara et al. [26]. However, relative low concentration of total phenols (9.245 mg CE/g) and flavonoids (2.14 mg QE/g) were found in G. lucidum sample from India [27]. Estimation of phenolic content is important because there are many reports that phenols and antioxidant activity are directly correlated [7,8]. Invariably, total phenolics content can be an indicator of the antioxidant capacity of the extract [28]. In addition, natural polyphenols have chainbreaking antioxidant activities and are believed to prevent many degenerative diseases, including cancer and atherosclerosis [29].

Table 1. Total phenolics and flavonoid contents of mushroom fruiting bodies

Mushroom extract	Total phenolics (mg GAE/g)	Total flavonoids (mg QE/g)
B. berkeleyi	32.99	22.40
G. lucidum	42.70	35.23

DPPH radical was used as a stable free radical to determine antioxidant activity. Antioxidants, on interaction with DPPH transfer an electron (hydrogen atom) to DPPH, neutralizing its free radical character [30]. Fig. 1 illustrates the concentration of DPPH radical due to the scavenging ability of extracts of B. berkeleyi and G. lucidum. Standard BHA was used as a reference. The percentage inhibition of the free radical was dose dependent. Increase in concentration gave corresponding increased % inhibition. In a concentration of 500 µg/ml of extracts and standard BHA, the descending order of DPPH radical scavenging is as follows: BHA (89.28%) > G. lucidum extract (86.28%) > B. berkeleyi extract (74.66%).

The DPPH radical scavenging capacity (IC_{50}) of the extracts is presented in Table 2. This is the concentration that decreases the initial DPPH radical concentration by 50% in each extract. The effectiveness of antioxidant properties is inversely correlated with IC50 values. Ganoderma lucidum methanol extract showed more potent antioxidant activity (IC₅₀, 14.6 µg/ml) than B. berkeleyi (IC50, 22.91µg/ml). However, BHA exhibited higher DPPH scavenging effect (IC₅₀, 4.82 µg/ml) than both extracts in the study. Rawat et al. [27] demonstrated that in DPPH assay, IC₅₀ value for methanolic extract of wild G. lucidum collected in India was 1.162 mg/ml 2, 2'-azino-bis[3-ethylbenzothiazoline-6and

sulphonic acid] (ABTS) assay, 0.555 mg/ml. *Ganoderma lucidum* extract in this study displayed better DPPH activity than this Indian sample.

Chelating activity of the mushroom extracts were also tested in a metal chelating assay. All the extracts showed metal chelating capacity in a concentration-dependent manner; B. berkeleyi (67.4%) and G. lucidum (82.6%) at 500 µg/ml whereas the standard ethylenediaminetetraacetic acid (EDTA) exhibited higher activity (Fig. 2). Nagaraj et al. [26] reported good metal chelating activity for methanol extract, followed by chloroform, aqueous and petroleum ether extracts of G. applanatum with IC₅₀ values, 50.84 µg/ml, 54.74 µg/ml, 58.13 µg/ml and 69.12 µg/ml respectively. Iron is an essential mineral for normal physiological activity of the human body, but excess can cause cellular damage and injury. The ferrous ions are the most effective pro oxidants in food systems, the good chelating effect would be beneficial and removal of free ion from circulation could be a promising approach to prevent oxidative stress induced disease [31]. Ferrous ion reacts with ferrozine forms violet Ferrozoin-Fe2+ color complex. Chelating compounds present in the extract prevent the formation of Ferrozoine-Fe²⁺ complex, which leads to decrease in the intensity of violet color [32].

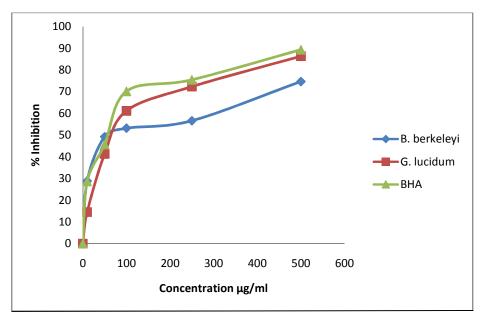


Fig. 1. Percentage DPPH scavenging activity of *B. berkeleyi* and *G. lucidum* extracts, and standard BHA

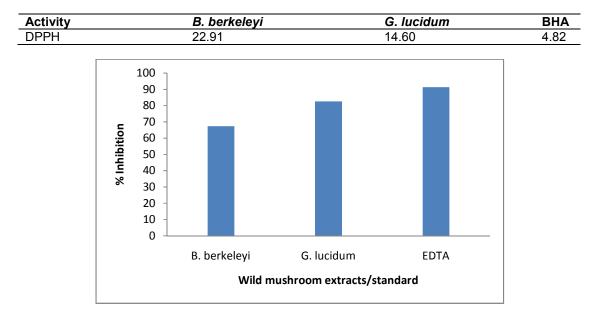
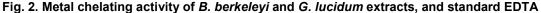


Table 2. IC₅₀ (µg/ mL) of *B. berkeleyi* and *G. lucidum* extracts



4. CONCLUSION

This study demonstrates that B. berkeleyi and G. lucidum extracts possess potent anti-oxidative properties by showing DPPH radical scavenging and metal chelating activities. This potent antioxidant activity may be attributed to its high phenolics and flavonoids content. The methanol extract can be considered as a valuable source of antioxidant products as the obtained values compared with the standard compounds. Furthermore, the results of this study substantiates the use of the extract of G. lucidum powder as an antioxidant agent and serves as a pointer for further exploitation of *B. berkelevi* mushroom for its antioxidant compounds and other potential biological activity.

CONSENT

It is not applicable.

ETHICAL APPROVAL

It is not applicable.

COMPETING INTERESTS

Authors have declared that no competing interests exist.

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